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4.8. The rate of the chemical reaction doubles for and increase of 10 K in absolute temperature from 298 K. Calculate E a. Ans.

4.9. The activation energy for the reaction,  $2\text{HI(g)} \rightarrow \text{H}_2 + \text{I}_2(\text{g})$  is 209.5 k J mol<sup>-1</sup> at 581 K. Calculate the fraction of molecules of reactants having energy equal to or greater than activation energy?

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